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Concept. 7. a) 13.6 mol

As b) 0.0732 mol Ne c)

1.12 mol Li d) $3.49 \times$

10^{-3} mol S 9. a) 35.4

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mol He b) 0.0606 mol

Ga c) 1.000 mol Mg d)

1.73 mol Cu

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The charge on an ion
is. Either positive or
negative. According to
the octet rule, a
calcium atom has a

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tendency to. Lose two
electrons.

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joecarter. 0. Save. Edit.
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. one atom of chlorine.
two atoms of chlorine.
one atom of oxygen.

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molar mass of C_6H_6O
= 94.11 g 21. a. molar
mass of SO_3 = 80.07 g
49.2 mg SO_3 1 g 1000
mg 1 mol 80.07 g =
6.14 10^{-4} mol SO_3 b.
molar mass of PbO_2 =
239.2 g 7.44 $\times 10^4$ kg
 PbO 2 1000 g 1 kg 1

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mol $239.2 \text{ g} = 3.11$

105 mol PbO_2 c. molar

mass of $\text{CHCl}_3 =$

119.37 g 59.1 g CHCl_3

$1 \text{ mol } 119.37 \text{ g} =$

0.495 mol CHCl_3 d.

molar mass of $\text{C}_2\text{H}_3\text{Cl}_3$

$= 133.39 \text{ g}$ $3.27 \dots$

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reactants. When a new substance is formed, the change is called a chemical ...

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