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Practice Problems In your notebook,

solve the following problems. ... Use the

Table 2.1 to answer the following

questions. 6.



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answer the following SECTION 7.1 IONS  
For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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The molecular structure of carbon dioxide is one carbon atom with two oxygen atoms on opposite sides of it.

\_\_\_\_\_ 13. Covalent bonds exist when combining atoms give up or accept electrons. \_\_\_\_\_ 14. A molecule contains

two atoms. Part C Matching Match each description in Column B to the correct

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term in Column A.

**05 CTR ch08 7/12/04 8:12 AM Page  
181 MOLECULAR COMPOUNDS 8**

680 Core Teaching Resources \_\_\_\_ 12.  
Radioactive decay a. is a spontaneous  
process. b. occurs when an unstable  
nucleus loses energy. c. does not require  
any input of energy. d. all of the above

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13. Which of the following types of radiation has no mass and no charge? a. alpha c. gamma b. beta d. positron \_\_\_\_\_
14. When U-238 undergoes radioactive decay by losing an alpha particle,

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(iii) the chemical symbol(s) for the most

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Write your answer in the space provided.  
Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas:

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Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.

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