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STUDY. PLAY. The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than one atom are called__

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monatomic ion. binary compounds.

nomenclature. oxyanions. an ion formed from a single atom. (remember to

change nonmetal.... compounds

composed of two different elements;

(compounds of 2.... naming system.

polyatomic ions that contain oxygen.

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thing merbobber hi hullo friends. Terms in this set (9) oxidation numbers rules.

1. the atoms in a pure element must equal zero

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Section 2 Review Flashcards ...

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1

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SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Section 1 Review Answers

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CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds SECTION 3
SHORT ANSWER Answer the following
questions in the space provided. 1. Label
each of the following statements as True
or False: True a. If the formula mass of

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one molecule is x amu, the molar mass is x g/mol. False b. Samples of equal numbers of moles of two different chemicals

7 Chemical Formulas and Chemical Compounds

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Modern Chemistry 35 Chapter Test
Name Class Date Chapter Test B,
continued Write the orbital notation for
the following elements in the space
provided Modern chemistry chapter 7
test answer key. 37. lithium, atomic
number 3 38. carbon, atomic number 6
39 Modern chemistry chapter 7 test
answer key. neon, atomic number 10

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PART VI Write the answers to the questions on the line to the left, and ...

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Holt McDougal Modern Chemistry 3
Chapter Test Chapter Test B, continued
16 Modern chemistry chapter 3 test b
answers. The measure of the ability of

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an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

Modern Chemistry Chapter 3 Test B Answers

Quiz Modern Chemistry 82 . Name

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Section Quiz, continued Class Date 6. If the amount of dissolved solute in a solution at a given temperature is greater than the amount that can permanently remain in solution at that temperature, the solution is said to be a. saturated. b. unsaturated.

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Chemical Formulas and Chemical
Compounds. CHAPTER 7 REVIEW
Chemical Formulas and Chemical
Compounds SECTION 4 SHORT ANSWER
Answer the following questions in the

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space provided.

Modern Chemistry Chapter 7 Review Answers Section 4

Preview this quiz on Quizizz. An ionic compound is NOT represented by a molecular formula because an ionic compound: Holt Modern Chemistry Chapter 6.3 Section Quiz DRAFT

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Name Class Date Chapter Test B,

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continued Refer to the following statement to answer questions 30-32: Dilute $\text{HCl}(\text{aq})$ and $\text{NaOH}(\text{aq})$ are mixed in chemically equivalent quantities. For instructors, they are a good resource for homework, quiz, or test questions, or practice for the AP Chemistry test. ...

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